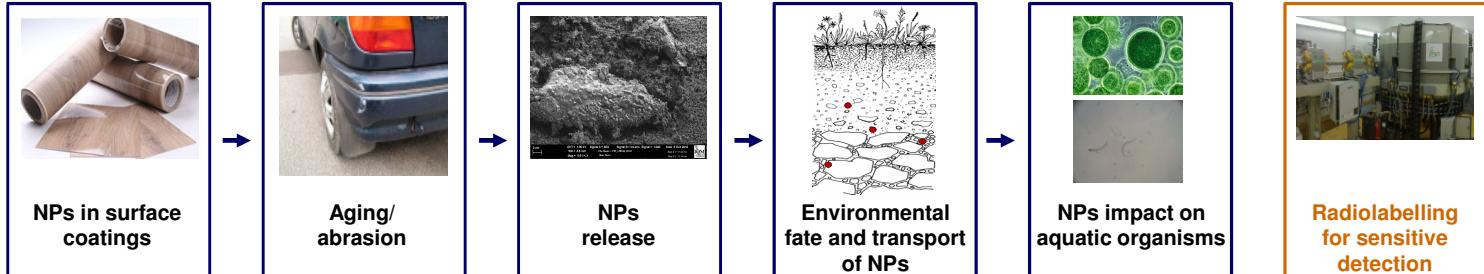


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## Motivation

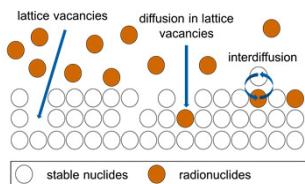


## Radiolabelling of engineered nanoparticles

### Isotopic radiolabelling of commercial TiO<sub>2</sub> and Ag<sup>0</sup> nanopowder

#### Labelling method:

→ Diffusion of radionuclides into NPs via low temperature annealing



#### Commercial radionuclides:

→ <sup>44</sup>Ti ( $T_{1/2} = 63$  a)  
→ <sup>110m</sup>Ag ( $T_{1/2} = 249.9$  d)



Target disk:  
Body: Al Ø 24 mm  
Target: Sc Ø 12 mm

#### Radiolabelling yield for Ag<sup>0</sup>: > 99 %

→ (0.2 – 1000) kBq <sup>110m</sup>Ag<sup>+</sup>/mg Ag<sup>0</sup> NPs

→ Low temperature annealing

T = 50°C – 105°C, t = 3 h

#### Radiolabelling yield for TiO<sub>2</sub>: > 99 %

→ (0.2 – 10) kBq <sup>44</sup>Ti<sup>+</sup>/mg TiO<sub>2</sub> NPs

→ Low temperature annealing

T = 180°C, t = 2 h

→ Up to 140 MBq <sup>45</sup>Ti<sup>+</sup>/mg TiO<sub>2</sub> NPs

→ Low temperature annealing

T = 180°C, t = 2 h

#### Specific activities and limits of detection

[<sup>110m</sup>Ag]Ag<sup>0</sup> 1 MBq/mg → 125 ng/L

[<sup>44</sup>Ti]TiO<sub>2</sub> 0.01 MBq/mg → 7.500 ng/L

(costly nuclide!)

[<sup>45</sup>Ti]TiO<sub>2</sub> 140 MBq/mg → 0.5 ng/L



#### TiO<sub>2</sub>-NP

→ <sup>48</sup>Ti(p,n)<sup>48</sup>V

<sup>48</sup>V:  $t_{1/2} = 15.97$  d / β<sup>+</sup> 100 %

Target system for irradiation of nanopowder samples



Target holder with activated TiO<sub>2</sub> NPs

### Proton activation of TiO<sub>2</sub> and Ag<sup>0</sup> nanopowder using a cyclotron

#### Ag<sup>0</sup>-NP

→ <sup>107</sup>Ag(p,3n)<sup>105</sup>Cd → <sup>105</sup>Ag

→ <sup>107</sup>Ag(p,p2n)<sup>105</sup>Ag

<sup>105</sup>Ag:  $t_{1/2} = 41.29$  d / β<sup>+</sup> 100 %

#### Specific activities and limits of detection

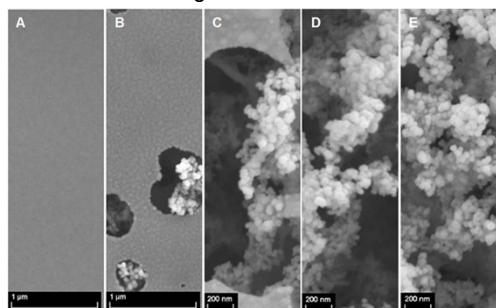
[<sup>105</sup>Ag]Ag<sup>0</sup> 69 kBq/mg → 1.8 µg/L

Scanditronix MC40, JRC

[<sup>48</sup>V]TiO<sub>2</sub> 36 MBq/mg → 2 ng/L

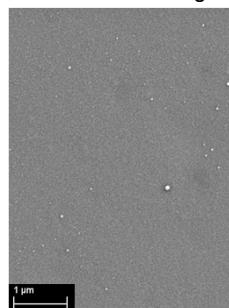
## Application of radiolabelled NPs for release studies

### labile surface coating → NP release is visible



Time-dependent degradation of a polyacrylate TiO<sub>2</sub> nanocomposite by UV-A-irradiation (intensity ~ 15 mW/cm<sup>2</sup>)  
 : A) original, B) after 2 days (50.000x); C) 4 days, D) 8 days, E) 16 days (75.000x); SEM micrographs: Leibniz Institute for Surface Modification.

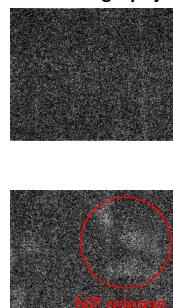
### stable surface coating



Degradation of a stable TiO<sub>2</sub> nanocomposite by UV-A-irradiation for 10 days (25.000x).

No visible NP release, but ~ 4 % of the applied TiO<sub>2</sub> is released and was only detected by measuring radioactivity!

### autoradiography



## References

### Isotopic radiolabelling of Ag<sup>0</sup>-NPs

Hildebrand & Franke *J Nanopart Res* 14 (2012)

### Proton activation of TiO<sub>2</sub>-NPs

Abbas K et al. *J Nanopart Res* 12 (2010)

Gibson N et al. *Arch Toxicol* 85 (2011)

Holzwarth U et al. *J Nanopart Res* 14 (2012)

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