Measurements of the Hygrscopicity of Fresh and Aged Nanoparticles for Assessing their Toxicity

S. Bezantakos¹ and G. Biskos^{2,3}



¹Université du Littoral Côte d'Opale, Maison de la Recherche en Environnement Industriel (MREI), 59140 Dunkerque, France

²Energy Environment and Water Research Center, The Cyprus Institute, Nicosia 2121, Cyprus

³Faculty of Civil Engineering and Geosciences, Delft University of Technology, Delft 2628-CN, The Netherlands



INTRODUCTION

The ability of aerosol nanoparticles (NPs) to take up water and increase in size (i.e., their hygroscopicity) can severely change their deposition characteristics in the human respiratory tract (Löndahl et al., 2007). The hygroscopicity of particles depends mainly on their size and chemical composition, which may differ between freshly emitted and aged nanoparticles. Hygroscopicity measurements conducted with the hygroscopic tandem differential mobility analyzer (HTDMA; Rader and McMurry,1986) are widely used for modeling the deposition characteristics of NPs in the human respiratory tract (e.g. Löndahl et al., 2008).

used in the emulsion. Interestingly the highest hygroscopicity is observed when the TD was operated at moderate temperatures, indicating that the hydrophobic part of the anticoagulating agents is more volatile.

• **Fig. 5:** Measured hygroscopicity of 20-nm (i.e., dry mobility diameter)



EXPERIMENTAL

In the frame of the CERASAFE project we measured the hygroscopic properties of fresh and aged antimony oxide (SbO₃/SbO₅) nanoparticles using an HTDMA system. In brief, an emulsion containing SbO₃/SbO₅ particles was aerosolized using an atomizer and their hygroscopicity was measured using the HTDMA system under two distinct conditions: i) without any additional treatment, and ii) when ozone was inserted in the system for simulating conditions of atmospheric aging. A thermal denuder (TD), able of reaching temperatures up to 380 °C was used upstream for evaporating any surfactants on the surface of the particles coming from the solution.

Initially, for corroborating that the anticoagulating agents in the emulsion of SbO₃/SbO₅ NPs were volatile, the size distribution of atomized and dried particles was measured at different temperature settings of the TD, using a Differential Mobility Analyzer (DMA; Knutson and Whitbey, 1975) and a Condensation Particle Counter (CPC; Agarwal & Sem, 1980; cf. fig 1). The SbO₃/SbO₅ hygroscopicity of nanoparticles was measured using the setup depicted in fig. 2 (i.e., the HTDMA) at three different temperature settings of the TD. The dN/dlogdp main operating principle of an HTDMA system is to size select particles with size narrow disutribution (i.e., monodisperse), using DMA-1. Particles are then exposed in well Nitrogen defined Relative (RH) Humidity conditions downstream DMA-1 by passing naffion through membrange humidifier their size and distribution is then measured by the second DMA (i.e., DMA-2) and the CPC. In this way the



Fig.1: Experimental setup for corroborating the volatility of the anticoagulating agents contained in the emulsion of SbO3/SbO5 nanoparticles.

Dried Aerosol Monodisperse Humidified Particles



Fig. 4: Measured HGFs of 20-nm aerosolized SbO₃/SbO₅ particles at various RHs and different temperature TD settings, without using Ozone.

presence of other hygroscopic mode(s), similarly to when externally mixed particles are measured with the HTDMA. Since in these series of experiments all NPs were introduced into the gas phase via atomization, an internally mixing state would be expected and indeed was observed during the hygroscopicity experiments without using ozone. This clearly indicates a chemical reaction, most probably including ozone and left overs of the anticoagulating agents, resulting in changing the size distribution of the monodisperse particles or even in forming new particles having different sizes than those selected by DMA-1 (i.e., dry diameter of 20 nm).

• **Fig. 6:** Experiments exposing dry (i.e., RH< 30%) and denuded 20-nm monodisperse NPs at 380 °C, only to ozone further corroborated this finding, as their size distributions were significantly broadened, resulting in bigger particles which explain well the observed very high HGFs measured in the experiments conducted with ozone.

3.5 []1.20	3500 []
------------	---------





Hygroscopic Growth Factor (HGF) of the sampled particles at variable RH conditions can be determined as: $HGF(RH) = d_m(RH)/d_m(dry)$, where $d_m(RH)$ and $d_m(dry)$ are the mobility diameters of the humidified and dry particles, respectively. This experimental setup offers also the ability of exposing the monodisperse particles (i.e., downstream DMA-1) to ozone for simulating atmospheric

CONCLUSIONS

- Anticoagulating agents used in emulsions containing nanoparticles affect their hygroscopicity.
- + The specific anticoagulating agents are volatile and can be partially evaporated using a thermal

aging conditions.

RESULTS

• **Fig. 3:** Measured size distributions of polydisperse SbO₃/SbO₅ nanoparticles contained in an emulsion together with anticoagulating agents at various TD temperatures. As the temperature in the TD stages is increasing, the particle number concentration decreases together with the spread of the distribution, indicating the evaporation of volatile matter.

• **Fig. 4:** Measured hygroscopicity of 20-nm (i.e., dry mobility diameter) SbO₃/SbO₅ nanoparticles at various RH conditions and TD temperature settings without ozone. The lowest hygroscopicity is exhibited when the TD is operated at the highest temperaures, corroborating that, at least part of the measured particle hygroscopicity can be attributted to the anticoagulating agents,



denuder.

+ Leftovers of the anticoagulating agents (i.e., downstream a TD at 380 °C) react with ozone to change the size of the existing NPs .

+ It is essential therfore to avoid the use of any chemical additives, either used as preservatives or anticoagulating agents when testing the physicochemical properties of nanoparticles.

+ This may be achieved by introducing the nanomaterial in the gas phase using other aerosolizing techniques, such as the spark abbletion method (Tabrizi et al., 2009).

+ Future work will be conducted using the spark abbletion method for assesing the hygroscopic properties of ceramic/metal oxide nanoparticles.

REFERENCES

Agarwal, J. K. and Sem, G. J. (1980), J. Aerosol Sci., 11:343–357. Knutson, E. O., and Whitby, K. T. (1975), J. Aerosol Sci., 6: 443–451. Löndahl J. et al., (2007), Inhalation Toxicology, 19:2, 109-116. Löndahl, J. et al., (2008), Inhalation Toxicology,20:10, 923-933. Rader, D. J. and McMurry P. H (1986).: J. Aerosol Sci. 17, 771–787 Tabrizi, N. S., Ullmann, M., Vons, V. A., Lafont, U., and Schmidt-Ott, A. (2009). J.

Nanopart. Res., 11:315–332.

Acknowledgement:



This research was conducted in the frame of CERASAFE project.