Project: VIGO



# Detection of reactive oxygen species in A549 cells

# DCF assay in A549 cells

AUTHORED BY:	DATE:
Cordula Hirsch	17.01.2014
REVIEWED BY:	DATE:
Harald Krug	09.04.2014
APPROVED BY:	DATE:

## **DOCUMENT HISTORY**

Effective Date	Date Revision Required	Supersedes		
15.02.2014	DD/MM/YYYY	DD/MM/YYYY		

Version	Approval Date	Desci	ription of the Change	Author / Changed by		
1.0	DD/MM/YYYY	All	Initial Document	Cordula Hirsch		

Document Type	Document ID	Version	Status	Page
SOP	O_DCF_A549	1.0		1/18

# **Table of Content**

1	Ir	ntrod	uction	. 3
2	Р	rincip	le of the Method	. 3
3	A	pplic	ability and Limitations	. 3
4	R	elate	d Documents	. 4
5	E	quipr	nent and Reagents	. 4
	5.1	E	quipment	. 4
	5.2	R	eagents	. 4
	5.3	R	eagent Preparation	. 5
	5	.3.1	Complete cell culture medium	. 5
	5	.3.2	1x concentrated Hank's Balanced Salt Solution (HBSS)	. 5
	5	.3.3	2x concentrated Hank's Balanced Salt Solution (HBSS)	. 6
	5	.3.4	Pluronic F-127	. 6
	5	.3.5	H₂DCF-DA	. 6
	5	.3.6	Deacetylation of H₂DCF-DA	. 6
	5	.3.7	Sin-1	. 7
6	Р	roced	lure	. 7
	6.1	G	eneral remarks	. 7
	6.2	Fl	ow chart	. 8
	6.3	C	ell seeding	. 8
	6	.3.1	Cell culture	. 8
	6	.3.2	Cell seeding into 96-well plate	. 8
	6.4	Р	rearrangements	. 9
	6	.4.1	Deacetylation of H₂DCF-DA	. 9
	6	.4.2	Dilution of nanomaterials	. 9
	6	.4.3	Dilution of Sin-1 (chemical positive control)	12
	6.5	Lo	oading of cells with H₂DCF-DA	12
	6.6	Р	reparation of cell-free control wells	12
	6.7	Α	pplication of stimuli and measurement	13
	6.8	D	ata evaluation	15
7	Q	uality	Control, Quality Assurance, Acceptance Criteria	15
8	Н	lealth	and Safety Warnings, Cautions and Waste Treatment	15
9	A	bbre	viations	16

Document ID

O\_DCF\_A549

Status

Page

2/18

Version

1.0

Document Type

SOP

10	References						
11	Annex A:	. 17					

## 1 Introduction

NMs can influence cellular systems in several ways. One very prominent and widely discussed cellular (as well as systemic) reaction towards NMs is the overproduction of reactive oxygen species (ROS) (see e.g. Donaldson et al. 2004; Johnston et al., 2010; Nel et al., 2006; Schins and Knaapen, 2007; Wiseman and Halliwell, 1996). In healthy cells ROS play an important role as messengers for both intra- and intercellular communication (e.g. Hancock, 2001; Held, 2010). Cells are generally used to handle such reactive species by anti-oxidant defense mechanisms that balance the absolute levels of ROS. Once these protective mechanisms fail, the increasing ROS levels lead to oxidative stress reactions. These include damage to DNA, proteins and lipids and can further lead to genotoxicity, inflammation and/or cell death. Thus, detecting elevated levels of ROS in cellular systems is an important tool to assess very early steps of potential cytotoxic effects.

## 2 Principle of the Method

The 2', 7'-dichlorodihydrofluorescein diacetate ( $H_2DCF$ -DA) assay is a widely used *in vitro* ROS-detection method. The non-fluorescent dye ( $H_2DCF$ -DA) is a chemically reduced form of fluorescein and cell-permeable. Intracellular esterases cleave off the diacetate (DA) moiety which renders the molecule ( $H_2DCF$ ) sensitive to oxidation by ROS. In its oxidized form dichlorofluorescein (DCF) is highly fluorescent and easily detectable e.g. using a fluorescent plate reader.

## 3 Applicability and Limitations

 $H_2DCF$  is a non-specific ROS detector sensitive to a number of different reactive species. This can be an advantage for initial screening approaches where neither the general existence nor the type of reactive species is known. This easy and fast screening assay yields **qualitative** results that serve as an indication for further investigations and is as such valuable.

As the  $H_2DCF$  molecule is not completely retained inside the cell the measured fluorescence values are thus the sum of intra- as well as extracellular ROS formation.

Non-specific esterases not only exist intracellularly but also in serum which is usually added to cell culture medium. To avoid dye cleavage even before cellular uptake it is important to perform the assay under **serum free conditions**, preferentially in Hank's balanced salt solution (HBSS).

**NM-related consideration**: The large (most often reactive) surface area of NMs may be able to process the H<sub>2</sub>DCF molecule to DCF without cellular contribution. This issue is addressed in this SOP in the cell free part of the 96-well plate. Furthermore NMs have been reported to interfere with fluorescence measurements by quenching an existing signal (for a review see Kroll et al., 2009). This issue is addressed in the related SOP "NM interference in the DCF assay". Both cell free controls

Document Type	Document ID	Version	Status	Page
SOP	O_DCF_A549	1.0		3/18

cannot be calculated against values from cellular measurements. They serve as qualitative estimations of NM only reactions that do not involve cellular contribution.

#### 4 Related Documents

**Table 1:** Documents needed to proceed according to this SOP and additional NM-related interference control protocols.

Document ID	Document Title
O_DCF_interference	NM interference in the DCF assay – Quenching effects – DCF
cell culture_A549	Culturing A549 cells
M_NM suspension_metal oxides	Suspending and diluting Nanomaterials – Metal oxides and NM purchased as monodisperse suspensions
M_NM suspension_ carbon based	Suspending and diluting Nanomaterials – Carbon based nanomaterials

## 5 Equipment and Reagents

#### 5.1 Equipment

- Centrifuge (for cell pelleting; able to run 15 ml as well as 50 ml tubes at 200 x g)
- Conical tubes (15 ml and 50 ml; polypropylene or polystyrene; e.g. from Falcon)
- Flat bottom 96-well cell culture plates
- Fluorescence reader for multi-well plates (to measure excitation/emission at wavelength maxima of: λex=485 nm and λem=528 nm)
- Hemocytometer
- Laminar flow cabinet (biological hazard standard)
- Light microscope (for cell counting and cell observation)
- Microreaction tubes (1.5 ml; e.g. from Eppendorf)
- Multichannel pipette (with at least 8 positions; volume range per pipetting step at least from 50  $\mu$ l to 200  $\mu$ l)
- Vortex<sup>®</sup>

#### 5.2 Reagents

For cell culturing:

- Fetal Calf Serum (FCS)
- L-glutamine
- Neomycin<sup>1)</sup>
- Penicillin<sup>1)</sup>
- Phosphate buffered saline (PBS)
- Roswell Park Memorial Institute medium (RPMI-1640)
- Streptomycin<sup>1)</sup>
- Trypsin-EDTA (0.05%)

SOP	O DCF A549	1.0	0 00.00	4/18
Document Type	Document ID	Version	Status	Page

<sup>1)</sup> bought as a 100x concentrated mixture of Penicillin, Streptomycin and Neomycin (PSN) e.g. from Gibco.

Buffers, solvents and detection dye itself:

- 2',7'-Dichlorofluorescin-diacetate (H<sub>2</sub>DCF-DA) [CAS number: 4091-99-0]
- 3-Morpholinosydnonimine hydrochloride (SIN-1)
- Calcium chloride dihydrate (CaCl<sub>2</sub>\*2 H<sub>2</sub>O) [CAS number: 10035-04-8]
- D-Glucose [CAS number: 50-99-7]
- Dimethyl sulfoxide (DMSO) [CAS number: 67-68-5]
- Disodium phosphate (Na<sub>2</sub>HPO<sub>4</sub>\*2H<sub>2</sub>O) [CAS number: 10028-24-7]
- Magnesium chloride hydrate (MgCl<sub>2</sub>\*6H<sub>2</sub>O) [CAS number: 7791-18-9]
- Magnesium sulfate heptahydrate (MgSO<sub>4</sub>\*7H<sub>2</sub>O) [CAS number: 7487-88-9]
- Methanol [CAS number: 67-56-1]
- Monosodium phosphate (NaH<sub>2</sub>PO<sub>4</sub>) [CAS number: 89140-32-9]
- Pluronic F-127 [CAS number: 9003-11-6]
- Potassium chloride (KCI) [CAS number: 7447-40-7]
- Potassium hydrogen phosphate (KH<sub>2</sub>PO<sub>4</sub>) [CAS number: 7778-77-0]
- Sodium chloride (NaCl) [CAS number: 8028-77-1]
- Sodium hydrogen carbonate (NaHCO<sub>3</sub>) [CAS number: 7542-12-3]
- Sodium hydroxide (NaOH) [CAS number: 1310-73-2]

Note: Corrosive! Handle with special care! (see chapter 8)

#### 5.3 Reagent Preparation

#### 5.3.1 Complete cell culture medium

Basic medium:

RPMI-1640

supplemented with:

- 10% FCS
- 1x PSN, which results in final concentrations of:
  - o 50 μg/ml Penicillin
  - o 50 μg/ml Streptomycin
  - o 100 μg/ml Neomycin
- 0.2 mg/ml L-glutamine

#### 5.3.2 1x concentrated Hank's Balanced Salt Solution (HBSS)

1 g/l D-glucose 185 mg/l CaCl<sub>2</sub> \* 2 H<sub>2</sub>O 400 mg/l KCl 60 mg/l KH<sub>2</sub>PO<sub>4</sub> 100 mg/l MgCl<sub>2</sub> \* 6 H<sub>2</sub>O 100 mg/l MgSO<sub>4</sub> \* 7 H<sub>2</sub>O

SOP	O DCF A549	1.0		5/18
Document Type	Document ID	Version	Status	Page

```
8 \text{ g/l} NaCl 350 \text{ mg/l} NaHCO<sub>3</sub> 60 \text{ mg/l} Na<sub>2</sub>HPO<sub>4</sub>*2 H<sub>2</sub>O
```

Dissolve all reagents in ddH<sub>2</sub>O and adjust the pH to 7.4. Store at 4°C.

#### 5.3.3 2x concentrated Hank's Balanced Salt Solution (HBSS)

Double-concentrated HBSS will be necessary to dilute nanomaterials (NM) for final application.

```
2 g/l D-glucose

370 mg/l CaCl<sub>2</sub> * 2 H<sub>2</sub>O

800 mg/l KCl

120 mg/l KH<sub>2</sub>PO<sub>4</sub>

200 mg/l MgCl<sub>2</sub> * 6 H<sub>2</sub>O

200 mg/l MgSO<sub>4</sub> * 7 H<sub>2</sub>O

16 g/l NaCl

700 mg/l NaHCO<sub>3</sub>

120 mg/l Na<sub>2</sub>HPO<sub>4</sub> *2 H<sub>2</sub>O
```

Dissolve all reagents in ddH<sub>2</sub>O and adjust the pH to 7.4. Store at 4°C.

#### **5.3.4 Pluronic F-127**

Stock:

• 160 ppm in ddH<sub>2</sub>O: 160 μg/ml (=16 mg/100 ml)

#### 5.3.5 H<sub>2</sub>DCF-DA

Stock:

5 mM in DMSO: 2.44 mg/ml

Working concentration:

• 50 μM in HBSS: 70 μl [5 mM] in 7 ml HBSS

#### 5.3.6 Deacetylation of H<sub>2</sub>DCF-DA

Prepare the following solutions freshly:

NaOH (0.01 M): 0.4 mg/ml NaOH

NaH<sub>2</sub>PO<sub>4</sub> (0.033 M): 5.2 mg/ml NaH<sub>2</sub>PO<sub>4</sub> \* 2 H<sub>2</sub>O adjust to pH 7.4

Preparation of 50 ml 50  $\mu$ M H<sub>2</sub>DCF:

Add 0.5 ml 5 mM  $H_2$ DCF-DA, 2.5 ml Methanol and 10 ml 0.01 M NaOH in a beaker. Stir for 30 minutes at room temperature (RT) in the dark (beaker covered in aluminum foil). Stop the reaction by adding 37.5 ml 33 mM Na $H_2$ PO $_4$ .

This solution is stable for two weeks in the dark at 4°C.

SOP	O DCF A549	1.0		6/18
Document Type	Document ID	Version	Status	Page

#### 5.3.7 Sin-1

Sin-1 is instable as soon as put in solution. Therefore work as fast as possible and prepare dilutions (compare 6.4.3) as freshly as possible.

#### Stock:

• 1 mM in HBSS: 0.21 mg/ml

If necessary this stock solution can be frozen in single use aliquots at -20°C.

## 6 Procedure

#### 6.1 General remarks

This SOP includes an optimized plate setup and dilution scheme to assess the oxidative potential of nanomaterials in A549 cells and in parallel in a cell free environment. Therefore the following plate layout is used:

		1	2	3	4	5	6	7	8	9	10	11	12
	Α		$\bigcirc$	$\bigcirc$	$\bigcirc$	$\bigcirc$	$\bigcirc$		$\bigcirc$	$\bigcirc$		$\bigcirc$	
ells	В												
A549 cells	С												
A54	D												
96	Е												
cell free	F												
89	G												
	Н												

Figure 1: General plate layout.

Cells are only seeded into wells B2-D11. Outermost wells A1-A12; A1-D1 and A12-D12 receive complete cell culture medium only. Wells E1 to H12 remain empty for the first 24 h and are then treated as described in the text below.

Document Type	Document ID	Version	Status	Page
SOP	O_DCF_A549	1.0		7/18

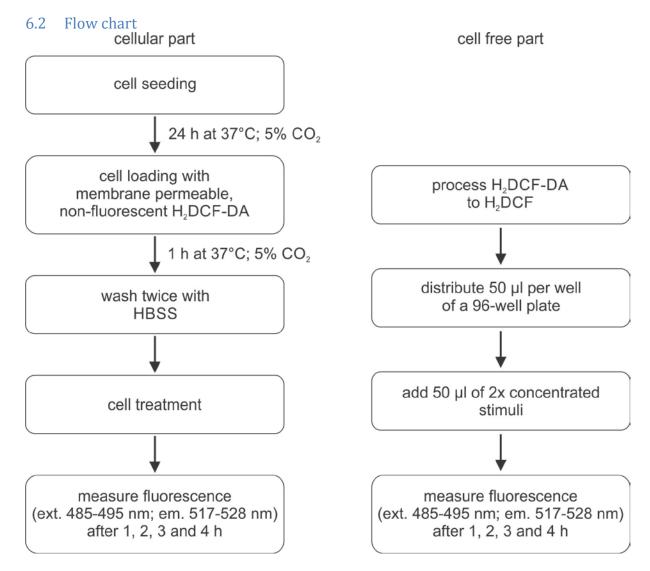


Figure 2: Brief outline of the workflow.

#### 6.3 Cell seeding

#### 6.3.1 Cell culture

A549 cells are grown in T75 cell culture flasks in a total volume of 20 ml of complete cell culture medium. They are kept at  $37^{\circ}$ C, 5% CO<sub>2</sub> in humidified air in an incubator (standard growth conditions according to SOP "Culturing A549 cells").

#### 6.3.2 Cell seeding into 96-well plate

- 24 hours prior to experimental start harvest and count cells as described in SOP "Culturing A549 cells".
- Seed  $2x10^4$  cells in 200  $\mu$ l complete cell culture medium per well into a 96-well cell culture plate
- For one 96-well plate (see Figure 3) 1x10<sup>6</sup> cells are suspended in 10 ml complete cell culture medium (1x10<sup>5</sup> cells/ml).
- Using a multichannel pipette (10 channels) 200  $\mu$ l of this cell suspension are distributed into each of the green wells (B2 to D11, Figure 3).

Document Type	Document ID	Version	Status	Page
SOP	O_DCF_A549	1.0		8/18

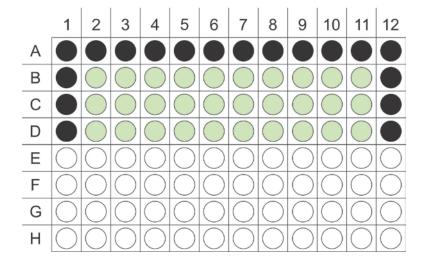


Figure 3: Cell seeding into a 96-well plate.

Cells are seeded at a density of  $2x10^4$  cells per well in 200  $\mu$ l complete cell culture medium into each of the green wells. Black wells receive 200  $\mu$ l complete cell culture medium each.

- Outermost wells (labeled in black in Figure 3) receive 200 μl complete cell culture medium only. These wells will serve as blank values later on and will be treated exactly the same as the cell containing wells (apart from not containing cells).
- Incubate cells overnight (24 hours) in a humidified incubator at standard growth conditions.

### 6.4 Prearrangements

#### 6.4.1 Deacetylation of H<sub>2</sub>DCF-DA

Perform deacetylation as described in 5.3.6. This can be done in advance as the H₂DCF is stable for two weeks at 4°C in the dark.

#### 6.4.2 Dilution of nanomaterials

For this SOP we distinguish two types of nanomaterials (NM) according to their solvent, suspension properties and highest concentrations used in the assay. See also respective related documents (3).

- (1) Metal oxide NM, Polystyrene beads and all NM delivered as monodisperse suspensions by the supplier: solvent either determined by the supplier or ddH<sub>2</sub>O; sub-diluted in ddH<sub>2</sub>O; highest concentration in assay 100 μg/ml
- (2) Carbon based NM: suspended and sub-diluted in 160 ppm Pluronic F-127; highest concentration in assay  $80 \,\mu\text{g/ml}$

Volumes given in the following dilution schemes are enough for one 96-well plate.

**Note:** "Mixing" in the context of diluting NMs means, the solvent containing tube is put on a continuously shaking Vortex® and the previous sub-dilution (or stock suspension, respectively) is put drop wise into the shaking solvent. The resulting suspension stays on the Vortex® for additional 3 seconds before proceeding with the next sub-dilution.

Document Type	Document ID	Version	Status	Page
SOP	O_DCF_A549	1.0		9/18

#### (1) Metal oxide NM:

Prepare serial sub-dilutions of the stock suspension (1 mg/ml) in ddH<sub>2</sub>O:

- Label seven microreaction tubes (1.5 ml total volume) with 1 to 7 (relates to steps 1-7 below).
- Add 400 μl of the 1 mg/ml stock suspension to tube 1.
- Add 200 μl ddH<sub>2</sub>O to tubes no. 2, 3, 5, 6 and 7.
- Add 240 µl ddH<sub>2</sub>O to tube 4.
- 1. 400  $\mu$ l NM stock suspension in ddH<sub>2</sub>O  $\rightarrow$  1 mg/ml (1)
- 2. 200  $\mu$ l of 1 mg/ml stock suspension (1) are mixed with 200  $\mu$ l of ddH<sub>2</sub>O  $\rightarrow$  500  $\mu$ g/ml (2)
- 3. 200  $\mu$ l of 500  $\mu$ g/ml (2) are mixed with 200  $\mu$ l ddH<sub>2</sub>O  $\rightarrow$  250  $\mu$ g/ml (3)
- 4. 160  $\mu$ l of 250  $\mu$ g/ml (3) are mixed with 240  $\mu$ l ddH<sub>2</sub>O  $\rightarrow$  100  $\mu$ g/ml (4)
- 5. 200  $\mu$ l of 100  $\mu$ g/ml (4) are mixed with 200  $\mu$ l ddH<sub>2</sub>O  $\rightarrow$  50  $\mu$ g/ml (5)
- 6. 200  $\mu$ l of 50  $\mu$ g/ml (5) are mixed with 200  $\mu$ l ddH<sub>2</sub>O  $\rightarrow$  25  $\mu$ g/ml (6)
- 7. 200  $\mu$ l ddH<sub>2</sub>O  $\rightarrow$  solvent control (7)

#### Preparation of final dilutions:

- Label seven microreaction tubes (1.5 ml) as follows:
  - 1. 200 μg/ml
  - 2. 100 μg/ml
  - 3. 50 μg/ml
  - 4. 20 μg/ml
  - 5. 10 μg/ml
  - 6.  $5 \mu g/ml$
  - 7. Solvent control: ddH<sub>2</sub>O
- Mix 3200  $\mu$ l 2x HBSS with 1920  $\mu$ l ddH<sub>2</sub>O. This mixture (A) is used in all following steps for the preparation of the final NM concentrations. Mixing NM sub-dilutions with (A) will result in 1x HBSS containing the respective NM concentrations.
- Add 640 μl (A) to each of the seven tubes.
- Mix on the Vortex® with 160  $\mu$ l of the respective NM sub-dilutions or the solvent (ddH<sub>2</sub>O):
  - 1. 160  $\mu$ l of the stock suspension (1 mg/ml) are mixed with 640  $\mu$ l (A)  $\rightarrow$  200  $\mu$ g/ml (1)
  - 2. 160  $\mu$ l of 500  $\mu$ g/ml sub-dilution are mixed with 640  $\mu$ l (A)  $\rightarrow$  100  $\mu$ g/ml (2)
  - 3. 160  $\mu$ l of 250  $\mu$ g/ml sub-dilution are mixed with 640  $\mu$ l (A)  $\rightarrow$  50  $\mu$ g/ml (3)
  - 4. 160  $\mu$ l of 100  $\mu$ g/ml sub-dilution are mixed with 640  $\mu$ l (A)  $\rightarrow$  20  $\mu$ g/ml (4)
  - 5. 160  $\mu$ l of 50  $\mu$ g/ml sub-dilution are mixed with 640  $\mu$ l (A)  $\rightarrow$  10  $\mu$ g/ml (5)
  - 6. 160  $\mu$ l of 25  $\mu$ g/ml sub-dilution are mixed with 640  $\mu$ l (A)  $\rightarrow$  5  $\mu$ g/ml (6)
  - 7. 160  $\mu$ l ddH<sub>2</sub>O are mixed with 640  $\mu$ l (A)  $\rightarrow$  solvent control (7)

SOP	O DCF A549	1.0		10/18
Document Type	Document ID	Version	Status	Page

#### (2) Carbon based NM:

Prepare serial sub-dilutions of the stock suspension (500 μg/ml) in 160 ppm Pluronic F-127:

- Label seven microreaction tubes (1.5 ml total volume) with 1 to 7 (relates to steps 1-7 below).
- Add 600 μl of the NM stock suspension in 160 ppm Pluronic F-127 to tube 1.
- Add 300 µl 160 ppm Pluronic F-127 totubes 2 to 7.
- 1. 600  $\mu$ l NM stock suspension in Pluronic F-127  $\rightarrow$  500  $\mu$ g/ml (1)
- 2. 300  $\mu$ l of the 500  $\mu$ g/ml stock suspension (1) are mixed with 300  $\mu$ l of Pluronic F-127  $\rightarrow$  250  $\mu$ g/ml (2)
- 3. 300  $\mu$ l of 250  $\mu$ g/ml (2) are mixed with 300  $\mu$ l Pluronic F-127  $\rightarrow$  125  $\mu$ g/ml (3)
- 4. 300  $\mu$ l of 125  $\mu$ g/ml (3) are mixed with 300  $\mu$ l Pluronic F-127  $\rightarrow$  62.5  $\mu$ g/ml (4)
- 5. 300  $\mu$ l of 62.5  $\mu$ g/ml (4) are mixed with 300  $\mu$ l Pluronic F-127  $\rightarrow$  31.25  $\mu$ g/ml (5)
- 6. 300  $\mu$ l of 31.25  $\mu$ g/ml (5) are mixed with 300  $\mu$ l Pluronic F-127  $\rightarrow$  15.63  $\mu$ g/ml (6)
- 7. 300  $\mu$ l 160 ppm Pluronic F-127  $\rightarrow$  solvent control (7)

#### Preparation of final dilutions:

- Label seven microreaction tubes (1.5 ml) as follows:
  - 1. 160 μg/ml
  - 2. 80 μg/ml
  - 3. 40 μg/ml
  - 4. 20 μg/ml
  - 5. 10 μg/ml
  - 6.  $5 \mu g/ml$
  - 7. Solvent control: Pluronic F-127
- Mix 3200  $\mu$ l 2x HBSS with 1152  $\mu$ l ddH<sub>2</sub>O. This mixture (B) is used in all following steps for the preparation of the final NM concentrations. Mixing NM sub-dilutions with (B) will result in 1x HBSS containing the respective NM concentrations.
- Add 544 μl (B) to each of the seven tubes.
- Mix on the Vortex® with 256 μl of the respective NM sub-dilutions or the solvent (160 ppm Pluronic F-127):
  - 1. 256  $\mu$ l of the stock suspension (500  $\mu$ g/ml) are mixed with 544  $\mu$ l (B)  $\rightarrow$  160  $\mu$ g/ml (1)
  - 2. 256  $\mu$ l of the 250  $\mu$ g/ml sub-dilution are mixed with 544  $\mu$ l (B)  $\rightarrow$  80  $\mu$ g/ml (2)
  - 3. 256  $\mu$ l of the 125  $\mu$ g/ml sub-dilution are mixed with 544  $\mu$ l (B)  $\rightarrow$  40  $\mu$ g/ml (3)
  - 4. 256  $\mu$ l of the 62.5  $\mu$ g/ml sub-dilution are mixed with 544  $\mu$ l (B)  $\rightarrow$  20  $\mu$ g/ml (4)
  - 5. 256  $\mu$ l of the 31.25  $\mu$ g/ml sub-dilution are mixed with 544  $\mu$ l (B)  $\rightarrow$  10  $\mu$ g/ml (5)
  - 6. 256  $\mu$ l of the 15.63  $\mu$ g/ml sub-dilution are mixed with 544  $\mu$ l (B)  $\rightarrow$  5  $\mu$ g/ml (6)
  - 7. 256  $\mu$ l 160 ppm Pluronic F-127 are mixed with 544  $\mu$ l (B)  $\rightarrow$  solvent control (7)

Document Type	Document ID	Version	Status	Page
SOP	O_DCF_A549	1.0		11/18

#### 6.4.3 Dilution of Sin-1 (chemical positive control)

Sin-1 is not stable in solution. Assure that the Sin-1 stock is thawed right before usage. Prepare the following dilutions as shortly before application (6.7) as possible. Tube labeling and HBSS distribution should be done in advance to speed up the final process of Sin-1 dilution and distribution into 96-well plate.

Prepare serial dilutions of the stock solution (1 mM) in HBSS. For one plate at least 800  $\mu$ l of each dilution and 1.5 ml of the stock solution are needed:

- Label five microreaction tubes (1.5 ml total volume) with 1 to 5 (relates to steps 1-5 below).
- Add 600 μl HBSS to tubes 1, 3 and 5.
- Add 1200 μl HBSS to tubes 2 and 4.

Shortly before usage finalize the dilution series as follows:

- 1. 600  $\mu$ l of 1 mM Sin-1 (stock solution) are mixed with 600  $\mu$ l HBSS  $\rightarrow$  500  $\mu$ M (1)
- 2.  $300 \mu l$  of 500  $\mu M$  Sin-1 (1) are mixed with 1200  $\mu l$  HBSS  $\rightarrow$  100  $\mu M$  (2)
- 3. 600  $\mu$ l of 100  $\mu$ M Sin-1 (2) are mixed with 600  $\mu$ l HBSS  $\rightarrow$  50  $\mu$ M (3)
- 4. 300  $\mu$ l of 50  $\mu$ M Sin-1 (3) are mixed with 1200  $\mu$ l HBSS  $\rightarrow$  10  $\mu$ M (4)
- 5. 600  $\mu$ l of 10  $\mu$ M Sin-1 (4) are mixed with 600  $\mu$ l HBSS  $\rightarrow$  5  $\mu$ M (5)

Application of NM as well as Sin-1 dilutions are shown in Figure 5 and described in 6.7 "Application of stimuli and measurement".

#### 6.5 Loading of cells with H<sub>2</sub>DCF-DA

• Prepare the 50  $\mu$ M working concentration of H<sub>2</sub>DCF-DA in HBSS. For one 96-well plate a final volume of 7 ml are needed:

#### 70 μl 5 mM H<sub>2</sub>DCF-DA stock + 7 ml HBSS

- Remove medium from wells A1 to D12 using a vacuum pump and a multichannel adapter. Be careful not to remove cells from the bottom of wells B2 to D11.
- Using a multichannel pipette (12 channels) add 100  $\mu$ l 50  $\mu$ M H<sub>2</sub>DCF-DA in HBSS per well (A1 to D12).
- Incubate plate in a humidified incubator at standard growth conditions for 60 minutes.
- Wash cells (wells A1 to D12) twice with pre-warmed (37°C) HBSS. Do not remove second HBSS solution before cell-free wells are prepared.

#### 6.6 Preparation of cell-free control wells

- Add 50  $\mu$ l ddH<sub>2</sub>O into each of the outermost wells (black wells in Figure 4, E1-H1; E12-H12; H2-H11).
- Add 50 μl deacetylated H<sub>2</sub>DCF (prepared according to 5.3.6) into each purple well (Figure 4, E2 to G11).

Document Type	Document ID	Version	Status	Page
SOP	O_DCF_A549	1.0		12/18

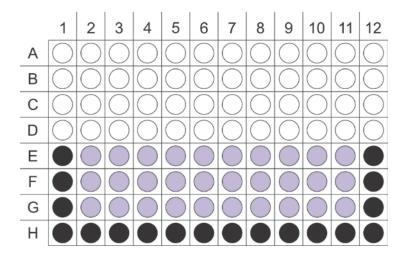


Figure 4: Cell free controls. Outermost wells (E1-H1; E12-H12; H1-H12) receive 50  $\mu$ l ddH<sub>2</sub>O each. Inner wells (E2-G11) receive 50  $\mu$ l deacetylated H<sub>2</sub>DCF.

## 6.7 Application of stimuli and measurement

Stick to the following chronological order.

**Note:** All NM dilutions have to be vortexed directly before application.

- 1. Make sure to have NM dilutions ready (6.4.2).
- 2. Prepare Sin-1 dilution (6.4.3).
- 3. Remove HBSS from wells A2 to D11. Leave HBSS in outer wells A1 to D1 and A12 to D12.
- 4. Add 100 μl of the respective NM dilution per well to wells A6 to D11 (shown in Figure 5 a).

wells	metal oxide NM concentration	carbon based NM concentration
A11-D11	100 μg/ml	80 μg/ml
A10-D10	50 μg/ml	40 μg/ml
A9-D9	20 μg/ml	20 μg/ml
A8-D8	10 μg/ml	10 μg/ml
A7-D7	5 μg/ml	5 μg/ml
A6-D6	solvent (ddH <sub>2</sub> O)	solvent (160 ppm Pluronic F-127)

5. Add **50**  $\mu$ I of the respective NM dilution per well to **cell free control** wells E6 to H11 (shown in Figure 5 b).

wells	metal oxide NM concentration	carbon based NM concentration
E11-H11	200 μg/ml	160 μg/ml
E10-H10	100 μg/ml	80 μg/ml
E9-H9	50 μg/ml	40 μg/ml
E8-H8	20 μg/ml	20 μg/ml
E7-H7	10 μg/ml	10 μg/ml
E6-H6	solvent (ddH₂O)	solvent (160 ppm Pluronic F-127)

**Note:** Applied concentrations are different for cellular and cell free wells! However, due to the 1:2 dilution of the NMs in the cell free wells (50  $\mu$ l H<sub>2</sub>DCF + 50  $\mu$ l NM dilution) final concentrations are the same.

Document Type	Document ID	Version	Status	Page
SOP	O_DCF_A549	1.0		13/18

6. Add 100 μl of respective Sin-1 dilution per well to wells A2 to D5 (shown in Figure 5 c).

wells	Sin-1 concentration
A5-D5	500 μM
A4-D4	50 μΜ
A3-D3	5 μΜ
A2-D2	solvent (HBSS)

7. Add **50**  $\mu$ l of the respective Sin-1 dilution per well to **cell free control** wells E2 to H11 (shown in Figure 5 d).

wells	Sin-1 concentration
E5-H5	1000 μΜ
E4-H4	100 μΜ
E3-H3	10 μΜ
E2-H2	solvent (HBSS)

**Note:** Applied concentrations are different for cellular and cell free wells! However, due to the 1:2 dilution of Sin-1 in the cell free wells (50  $\mu$ l H<sub>2</sub>DCF + 50  $\mu$ l Sin-1 dilutions) final concentrations are the same.

- 8. Incubate plate in a humidified incubator at standard growth conditions.
- 9. Measure fluorescence in a multi-well plate reader after 1, 2, 3 and 4 hours. After each measurement place plate back into incubator.

Fluorescence settings: excitation at  $\lambda$ =485-495 nm, emission at  $\lambda$ =517-528 nm.

Document Type	Document ID	Version	Status	Page
SOP	O_DCF_A549	1.0		14/18

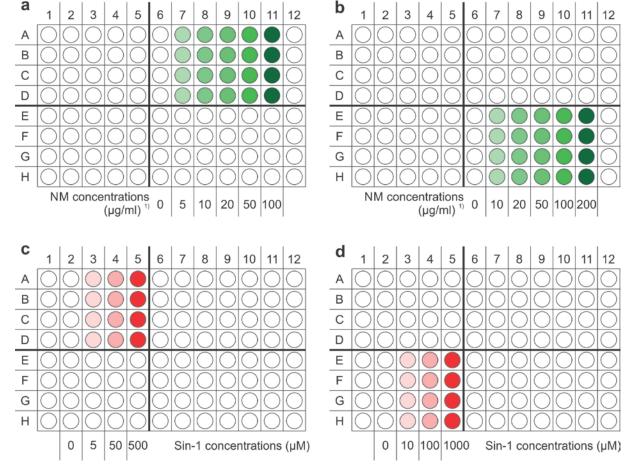


Figure 5: Application of stimuli.

a) Application of NM to cells first (wells A6-D11). b) Secondly, apply NM to cell free wells (E6-H11). c) Subsequently add Sin-1 to cells (A2-D5) and d) finally also to cell free wells (E2-H5).

#### 6.8 Data evaluation

Data are presented as blank corrected fluorescence values and represent the mean of three technical replicates. To calculate this, blank values (wells A2 to A11 for the cellular part and wells H2 to H11 for the cell free part) are subtracted from the respective sample values. For example: value in well A11 is subtracted from value in well B11, C11 and D11. B11\*=B11-A11; C11\*=C11-A11; D11\*=D11-A11. The mean and standard deviation is then calculated from the resulting blank corrected values B11\*, C11\* and D11\*.

## 7 Quality Control, Quality Assurance, Acceptance Criteria

## 8 Health and Safety Warnings, Cautions and Waste Treatment

Cell seeding has to be carried out under sterile conditions in a laminar flow cabinet (biological hazard standard). For this only sterile equipment must be used and operators should wear laboratory coat and gloves (according to laboratory internal standards).

Document Type	Document ID	Version	Status	Page
SOP	O_DCF_A549	1.0		15/18

<sup>&</sup>lt;sup>1)</sup> NM concentrations given here refer to metal oxide NM. Carbon based NM concentrations are detailed in the text.

Discard all materials used to handle cells (including remaining cells themselves) according to the appropriate procedure for special biological waste (i.e. by autoclaving).

NaOH is **corrosive**. It causes severe burns. Wear especially eye/face protection.

#### 9 Abbreviations

DA diacetate

DCF 2', 7'-Dichlorofluorescein ddH₂O double-distilled water DMSO Dimethyl sulfoxide

EDTA Ethylenediaminetetraacetic acid

FCS fetal calf serum

g constant of gravitation

H<sub>2</sub>DCF 2', 7'-Dichlorodihydrofluorescein

H<sub>2</sub>DCF-DA 2', 7'-Dichlorodihydrofluorescein-diacetate

HBSS Hank's balanced salt solution

NM nanomaterial

PBS phosphate buffered saline

ppm parts per million

PSN Penicillin, Streptomycin, Neomycin

ROS Reactive oxygen species

RPMI Roswell Park Memorial Institute medium

RT room temperature

Sin-1 3-Morpholinosydnonimine

#### 10 References

Donaldson K, Stone V, Tran CL, Kreyling W, Borm PJ; 2004: Nanotoxicology. *Occup. Environ. Med.* 61(9): 727-728

Hancock JT, Desikan R, Neill SJ; 2001: Role of reactive oxygen species in cell signaling pathways. *Biochemical and Biomedical Aspects of Oxidative Modification* 29(2): 345-350

Held P; 2010: An Introduction to reactive oxygen species. BioTek White Paper: Winooski, VT, USA.

Johnston HJ, Hutchison GR, Christensen FM, Peters S, Hankin S, Aschberger K, Stone V; 2010: A critical review of the biological mechanisms underlying the in vivo and in vitro toxicity of carbon nanotubes: The contribution of physic-chemical characteristics. *Nanotoxicology* 4(2): 207-246

Kroll A, Pillukat MH, Hahn D, Schnekenburger J; 2009: Current *in vitro* methods in nanoparticle risk assessment: Limitations and challenges. *Eur. J. Pharm. Biopharm.* 72(2): 370-377

Nel A, Xia T, Mädler L, Li N; 2006: Toxic potential of materials at the nanolevel. *Science* 311(5761): 622-627

Schins RP, Knaapen AM; 2007: Genotoxicity of poorly soluble particles. Inhal. Toxicol. 19: 189-198

Wiseman H, Halliwell B; 1996: Damage to DNA by reactive oxygen and nitrogen species: role in inflammatory disease and progression to cancer. *Biochem. J.* 313: 17-29

Document Type	Document ID	Version	Status	Page
SOP	O_DCF_A549	1.0		16/18

## 11 Annex A:

#### Additional background information:

As  $H_2DCF$  should be retained inside the cell the  $H_2DCF$ -DA method has been reported to specifically detect intracellular ROS production (see e.g. Kohno, 2010; Wardman, 2008). However, some leakage of the dye to the extracellular space is possible (Tarpey, et al., 2004). Reactive species outside the cell can then process the  $H_2DCF$ -molecule. The resulting fluorescence values are thus the sum of intra- as well as extracellular ROS formation.

Measurements cannot be performed **quantitatively**: ROS – as implicated in the name – are highly reactive and short lived species. Therefore it is important to measure as shortly after ROS induction as possible. This is achieved by loading the  $H_2DCF$ -DA dye prior to cell treatment. With the processed dye ( $H_2DCF$ ) in place any ROS pulse can be directly measured. However finding a "stable" positive reaction control for this assay is difficult. Any chemical used has to be as reactive as naturally occurring ROS and as a consequence cannot be stable. 3-Morpholinosydnonimine (Sin-1) is frequently used as the chemical positive control for the  $H_2DCF$ -DA assay (see e.g.: Buerki-Thurnherr et al., 2012; Limbach et al., 2007; Lipton et al., 1993; Piret et al., 2013; Wang et al., 1999). It works well to qualitatively see, if the assay per se worked or not. However, absolute values vary strongly due to the reactive and thus instable nature of the molecule. This is the case not only for Sin-1 but for any ROS-inducing compound. Prolonged handling steps during e.g. Sin-1 dilution or repeated freeze-thaw cycles (which have to be avoided!) reduce the activity of the molecule and with it also fluorescence values. Quantitatively comparing results from one experiment to another – or from one lab to another – is therefore difficult to impossible (Roesslein et al., 2013).

#### **References in Annex A:**

Buerki-Thurnherr T, Xiao L, Diener L, Arslan O, Hirsch C, Maeder-Althaus M, Grieder K, Wampfler B, Mathur S, Wick P, Krug HF; 2012: *In vitro* mechanistic study towards a better understanding of ZnO nanoparticle toxicity. *Nanotoxicology* 7(4): 402-416

Kohno M; 2010: Applications of electron spin resonance spectrometry for reactive oxygen species and reactive nitrogen species research. *J. Clin. Biochem. Nutr.* 47: 1-11

Limbach LK, Wick P, Manser P, Grass RN, Bruinink A, Stark WJ; 2007: Exposure of engineered nanoparticles to human lung epithelial cells: Influence of chemical composition and catalytic activity on oxidative stress. *Environ. Sci. Technol.* 41: 4158-4163

Lipton SA, Choi YB, Pan ZH, Lei SZ, Chen HS, Sucher NJ, Loscalzo J, Singel DJ, Stamler JS; 1993: A redox-based mechanism for the neuroprotective and neurodestructive effects of nitric oxide and related nitroso-compounds. *Nature* 364: 626-632

Piret JP, Jacques D, Audinot JN, Mejia J, Boilan E, Noel F, Fransolet M, Demazy C, Lucas S, Saout C, et al.; 2012: Copper(II) oxide nanoparticles penetrate into HepG2 cells, exert cytotoxicity via oxidative stress and induce pro-inflammatory response. *Nanoscale* 4: 7168-7184

Document Type	Document ID	Version	Status	Page
SOP	O_DCF_A549	1.0		17/18

Roesslein M, Hirsch C, Kaiser JP, Krug HF, Wick P; 2013: Comparability of *in vitro* tests for bioactive nanoparticles: A common assay to detect reactive oxygen species as an example. *Int. J. Mol.Sci.* 14: 24320-24337

Tarpey MM, Wink DA, Grisham MB; 2004: Methods for detection of reactive metabolites of oxygen and nitrogen: *in vitro* and *in vivo* considerations. *Am. J. Physiol. Regul. Integr. Comp. Physiol.* 286: 431-44

Wang H, Joseph JA; 1999: Quantifying cellular oxidative stress by dichlorofluorescein assay using microplate reader. *Free Radic. Biol. Med.* 27: 612-616

Wardman P; 2008: Methods to measure the reactivity of peroxynitrite-derived oxidants toward reduced fluoresceins and rhodamines. *Methods. Enzymol.* 411: 261-282

SOP	O_DCF_A549	1.0		18/18
Document Type	Document ID	Version	Status	Page